

Thin Layers of Organic Semiconductors for Chemical Sensors



1. Introduction

Phthalocyanines (porphyrine derivatives, Pcs) belong to a group of compounds which are reasonably utilized in gas/liquid analysis, because the interaction with a variety of different compounds (NO_x, Cl₂, O₃, H₂S, NH₃, VOCs, ...) results in significant **change of resistance**. 1,2

The **aim of diploma thesis** was twofold:

- 1. to achieve effective adjustment of electrical resistance with nanostructuring,
- 2. to enable detection of markers of explosives with chemiresistors based on Pcs.

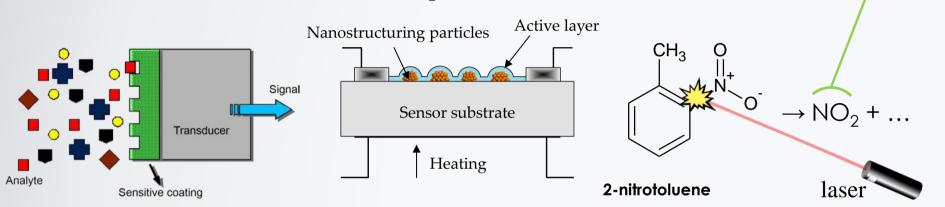


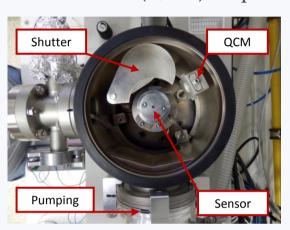
Fig. 1: General sensor scheme.³

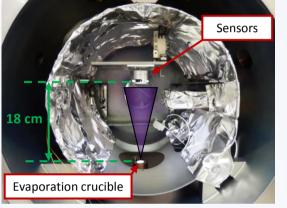
Fig. 2: Cross-section sensor structure. Fig. 3: Indirect detection of taggants.

2. Experimental

Metallic nanostructures (Au, Pd) on the alumina surface were prepared by DC magnetron sputtering and in-situ annealing. Formation of metallic nanoparticles was inspected using scanning electron microscopy and continual resistance measurements.

Silver phthalocyanine active layers were prepared by Organic Molecular Evaporation (OME) method under high vacuum conditions. The thickness was measured with Quartz Crystal Microbalance (QCM). Vapours of taggants were exposed to UV light generated in a Nd-YAG laser.





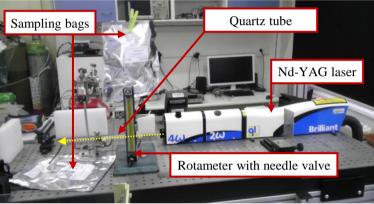
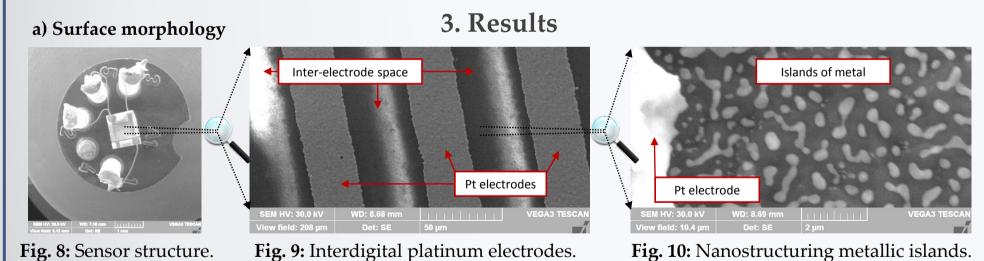


Fig. 5: Sputtering chamber.

Fig. 6: OME apparatus.

Fig. 7: Laser exposition apparatus.



b) Indirect detection of taggants 1E+10 vapours of 2-NT (150 ppr Gradual desorption of nitrogen **G** 1E+09 Selective sorption of NO₂ Response with magnitude of 373 dioxide from AgPc layer ≥ 1E+08 Potential for detection of tens of ppb 1E+07 10 20 $t \, (\min)$

4. Conclusions

To conclude, this contribution employs photo-induced decomposition of nitro compounds for their indirect detection by means of NO₂ sensors. The created methodology has a potential to detect nitro compounds which are used as markers of explosives (2-nitrotoluene, 4-nitrotoluene, 2,4dinitrotoluene, 2,3-dimethyl-2,3-dinitrobutane, ...) in concentrations as low as tens of ppb. Application of NO₂ sensors instead of expensive analytical equipment (ion mobility spectrometers, gas chromatographs ...) promises significant cost savings in detections of explosives.

Moreover, metal nanoparticles are used to decrease electrical resistance (up to 100x) of sensors based on phthalocyanines. This reduces the requirements for the level of attached ohmmeter device and as a result it contributes to another reduction of costs.

5. References

- 1.P. Fitl, M. Vrnata, D. Kopecky, J. Vlcek, J. Skodova, J. Bulir, M. Novotny and P. Pokorny, Applied Surface Science, 2014, **302**, 37-41
- 2.R. Zhou, F. Josse, W. Gopel, Z. Z. Ozturk and O. Bekaroglu, Appl Organomet Chem, 1996, 10, 557-577.
- 3.Z. Z. Öztürk, N. Kılınç, D. Atilla, A. G. Gürek and V. Ahsen, J Porphyr Phthalocya, 2009, 13, 1179-1187.









